

Directions:

- Using your common ions, write neutral compounds for each problem.
- Use subscripts to indicate more than one atom within a compound.
- SHOW YOUR WORK!!!!!
 - This includes: Symbols for each ion including charges, CROSSING OVER ARROWS, REDUCING TO LOWEST TERMS, and a <u>rewritten</u> final answer with a **BOX** around it!

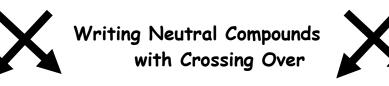
1	Potassium Bromide	6	Aluminum Carbonate		
2	Calcium Fluoride	7	Manganese (IV) Oxide		
3	Copper (II) Bromide	8	Calcium Carbonate		
4	Ammonium Carbonate	9	Antimony (III) Phosphate		
5	Aluminum Cyanide	10	Make up your own!!! Write the name out and then show how you would go from the name to the neutral formula.		



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